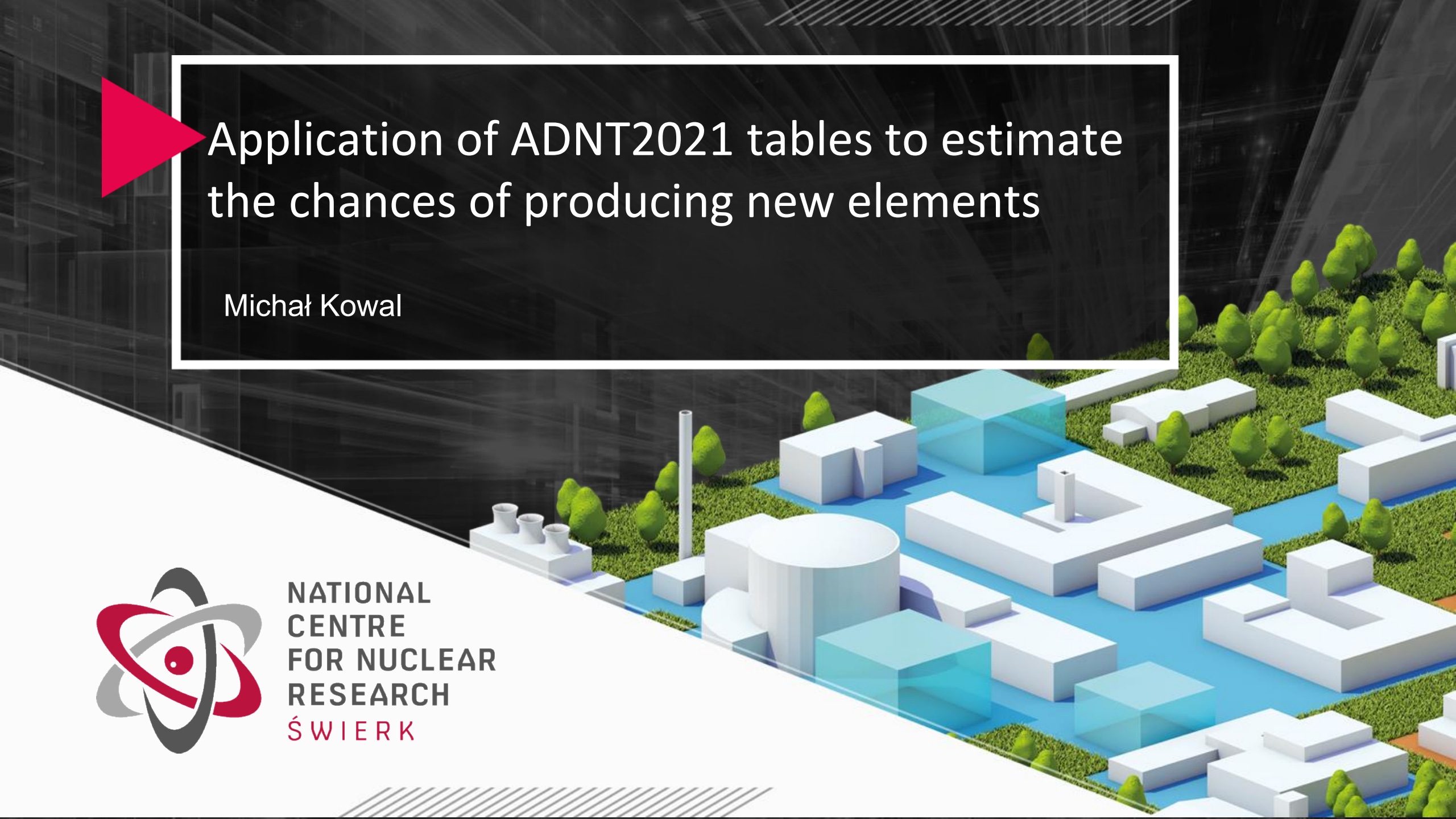




Application of ADNT2021 tables to estimate the chances of producing new elements

Michał Kowal



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Properties of heaviest nuclei with $98 \leq Z \leq 126$ and $134 \leq N \leq 192$

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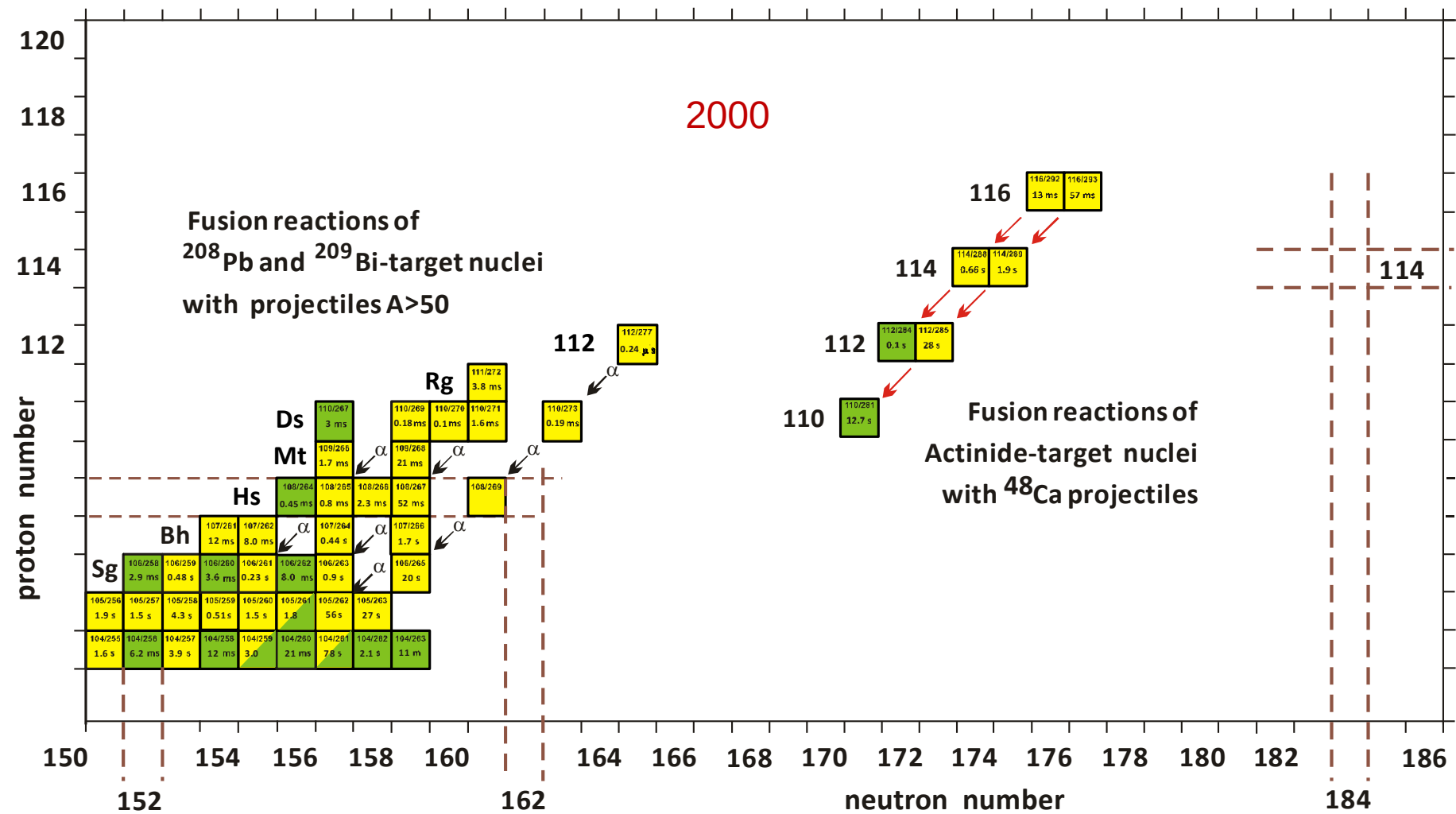
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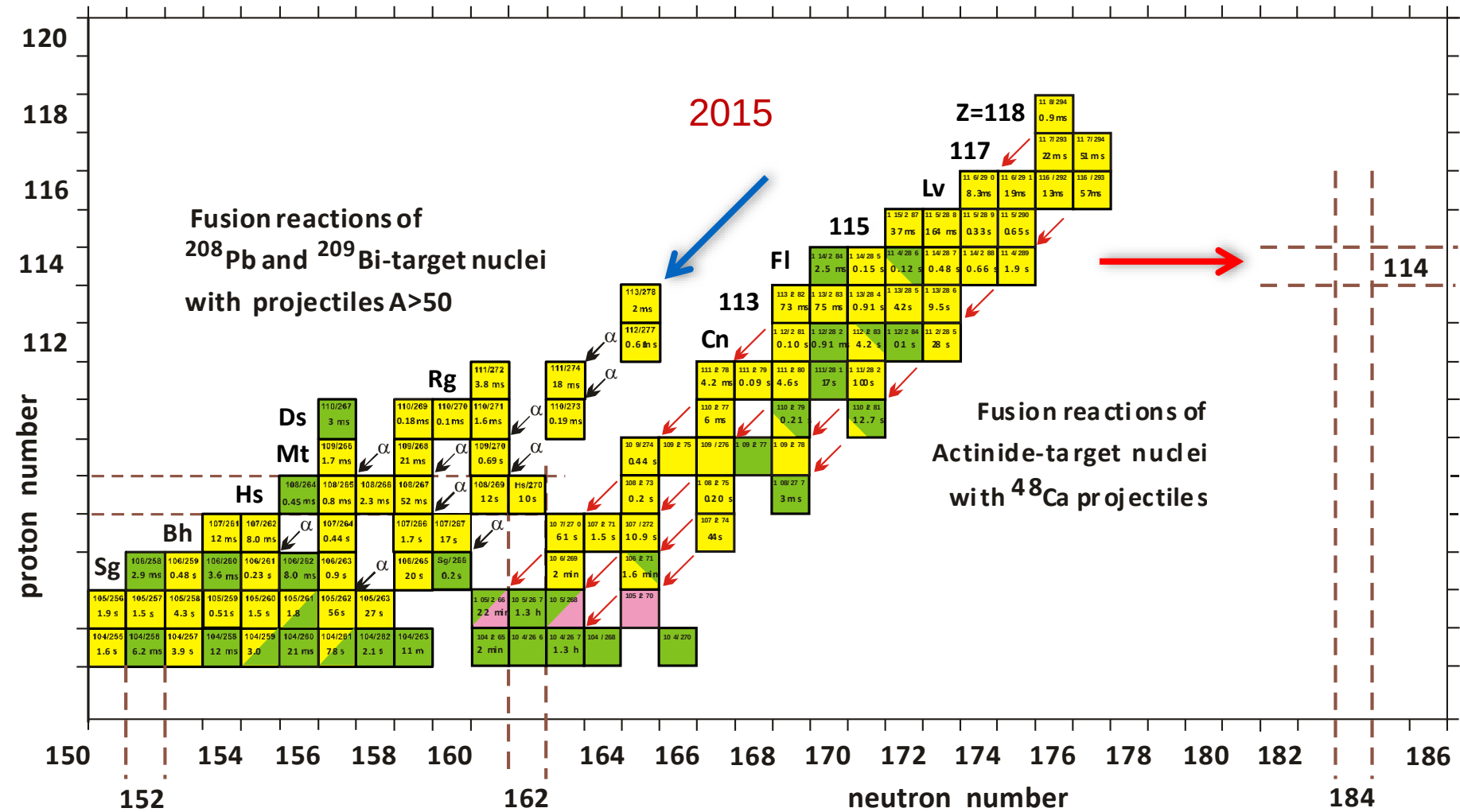
ABSTRACT

We systematically determine ground-state and saddle-point shapes and masses for 1305 heavy and superheavy nuclei with $Z = 98\text{--}126$ and $N = 134\text{--}192$, including odd- A and odd-odd systems. From these we derive static fission barrier heights, one- and two-nucleon separation energies, and Q_α values for g.s. to g.s. transitions. Our study is performed within the microscopic-macroscopic method with the deformed Woods-Saxon single-particle potential and the Yukawa-plus-exponential macroscopic energy taken as the smooth part. We use parameters of the model that were fitted previously to masses of even-even heavy nuclei. For systems with odd numbers of protons, neutrons, or both, we use a standard BCS method with blocking. Ground-state shapes and energies are found by the minimization over seven axially-symmetric deformations. A search for saddle-points was performed by using the "imaginary water flow" method in three consecutive stages, using five- (for nonaxial shapes) and seven-dimensional (for reflection-asymmetric shapes) deformation spaces. Calculated ground-state mass excess, nucleon separation- and Q_α energies, total, macroscopic (normalized to the macroscopic energy at the spherical shape) and shell corrections energies, and deformations are given for each nucleus in Table 1. Table 2 contains calculated properties of the saddle-point configurations and the fission barrier heights. In Tables 3-7, are given calculated ground-state, inner and outer saddle-point and superdeformed secondary minima characteristics for 75 actinide nuclei, from Ac to Cf, for which experimental estimates of fission barrier heights are known. These results are an additional test of our model.

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Borrowed from Yu.Ts. Oganessian.



SYNTHESIS SCENARIOS

COLD ($102 < Z < 113$)

- the strongly bound target nuclei (^{208}Pb or ^{209}Bi) are bombarded with projectiles ranging from Ca to Zn;
- the excitation energy of the resulting compound nucleus is usually in the range of 10 to 20 MeV;
- as the target-projectile symmetry increases, the compound nucleus production cross section decreases.

HOT ($112 < Z < 118$)

- the deformed actinide target-nuclei (from U to Cm) are bombarded with a doubly magic ^{48}Ca projectile;
- the excitation energy of the resulting compound nucleus is usually in the range of 30 to 40 MeV, and the dominant evaporation channels are 3n and 4n channels;
- the evaporation residue cross sections do not show any strong dependence on the target-projectile symmetry and are at the picobarn level.

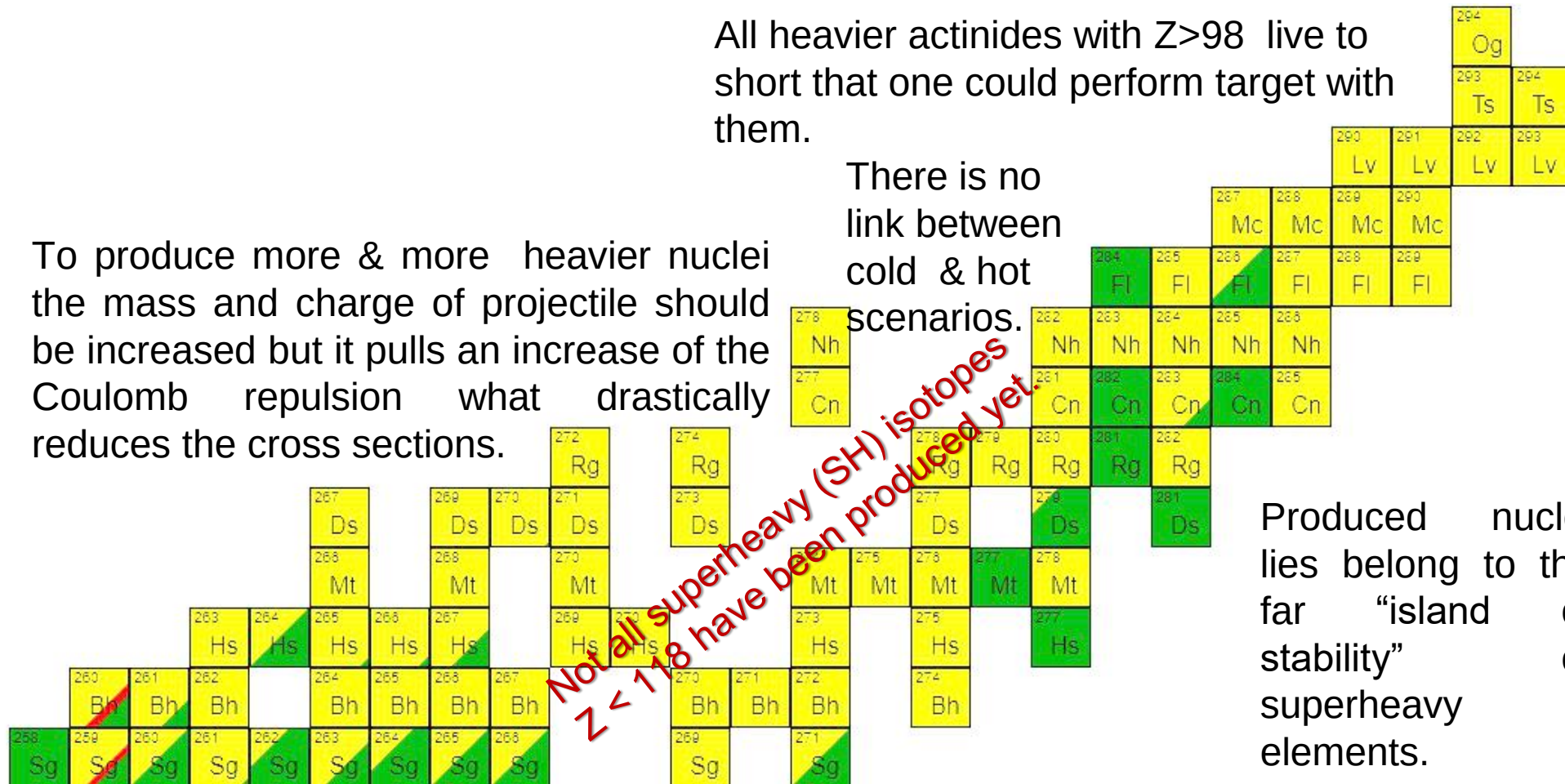
the issues

Attempts of going beyond the reactions Act. + ^{48}Ca by using heavier projectiles like ^{50}Ti , ^{54}Cr , ^{58}Fe , and ^{64}Ni gave no results so far.

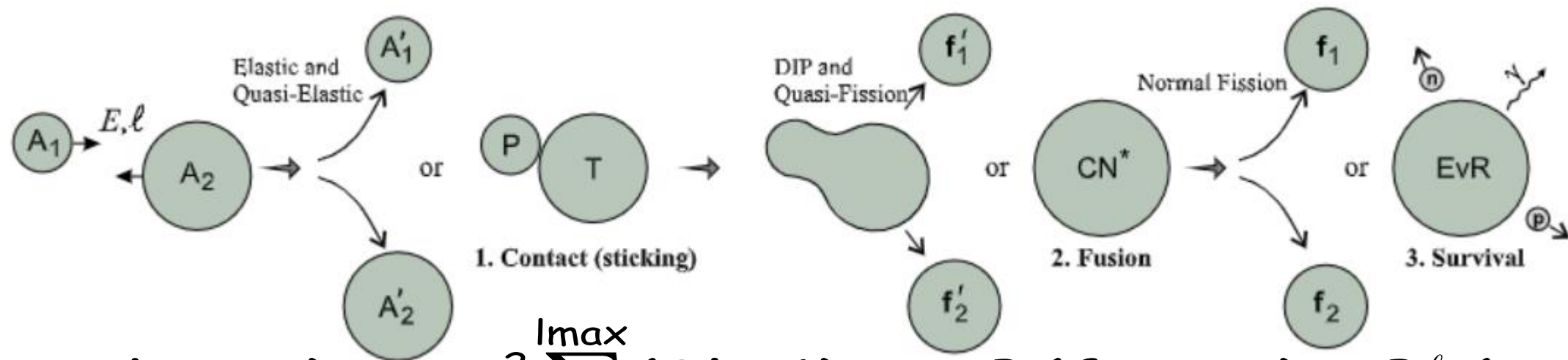
All heavier actinides with $Z > 98$ live so short that one could perform target with them.

There is no link between cold & hot scenarios.

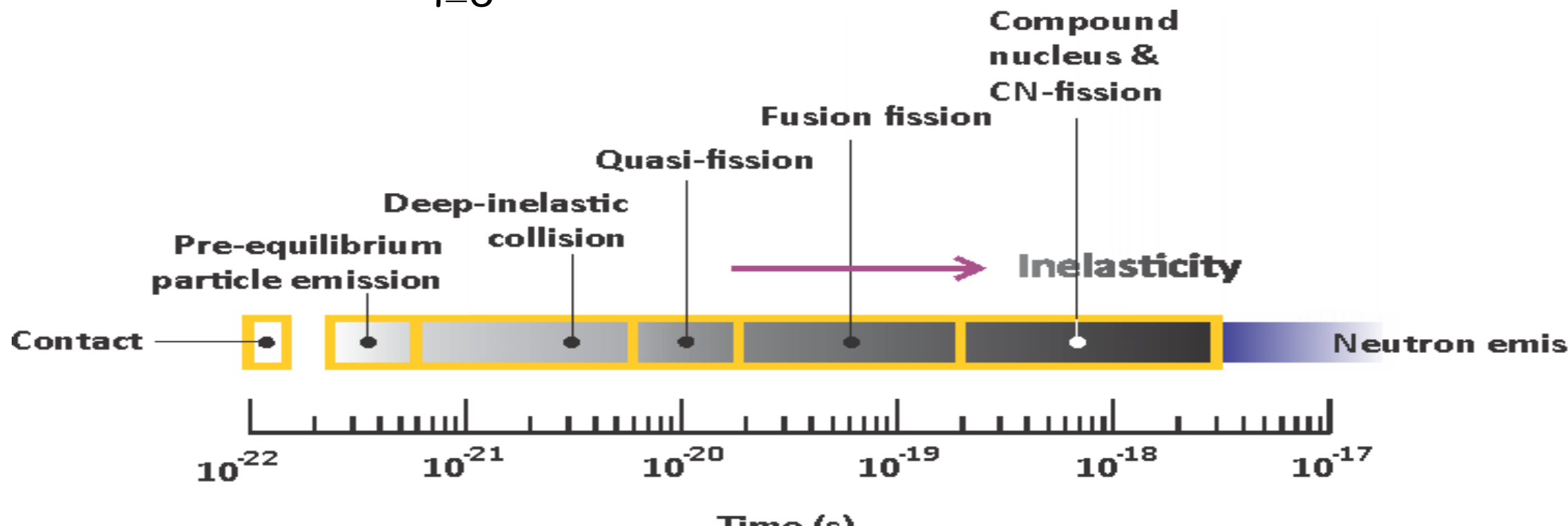
To produce more & more heavier nuclei the mass and charge of projectile should be increased but it pulls an increase of the Coulomb repulsion what drastically reduces the cross sections.

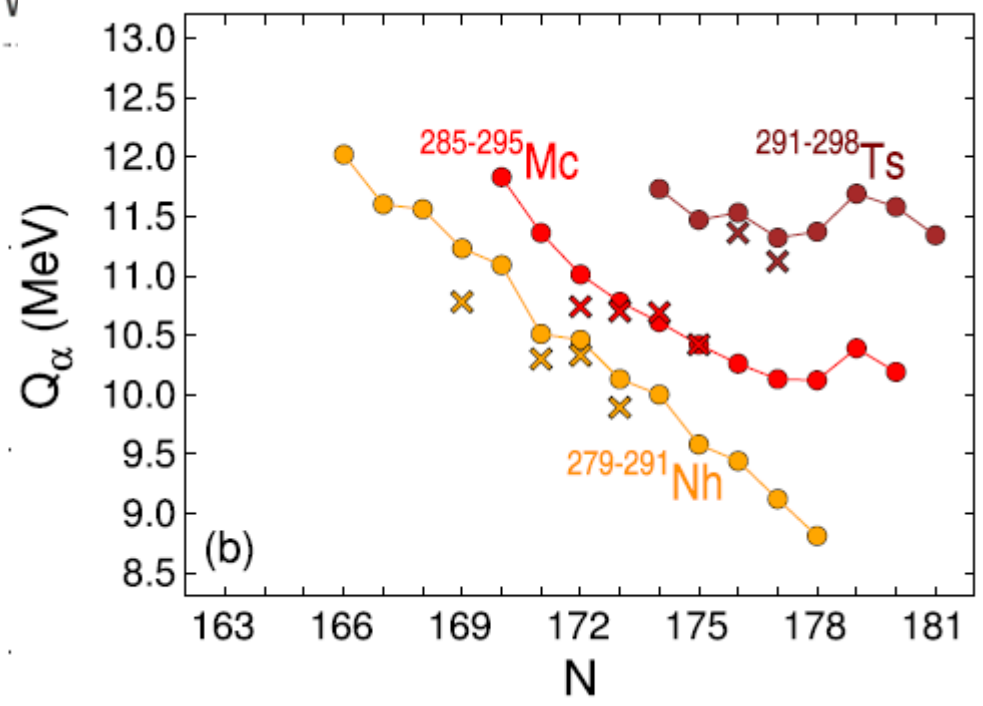
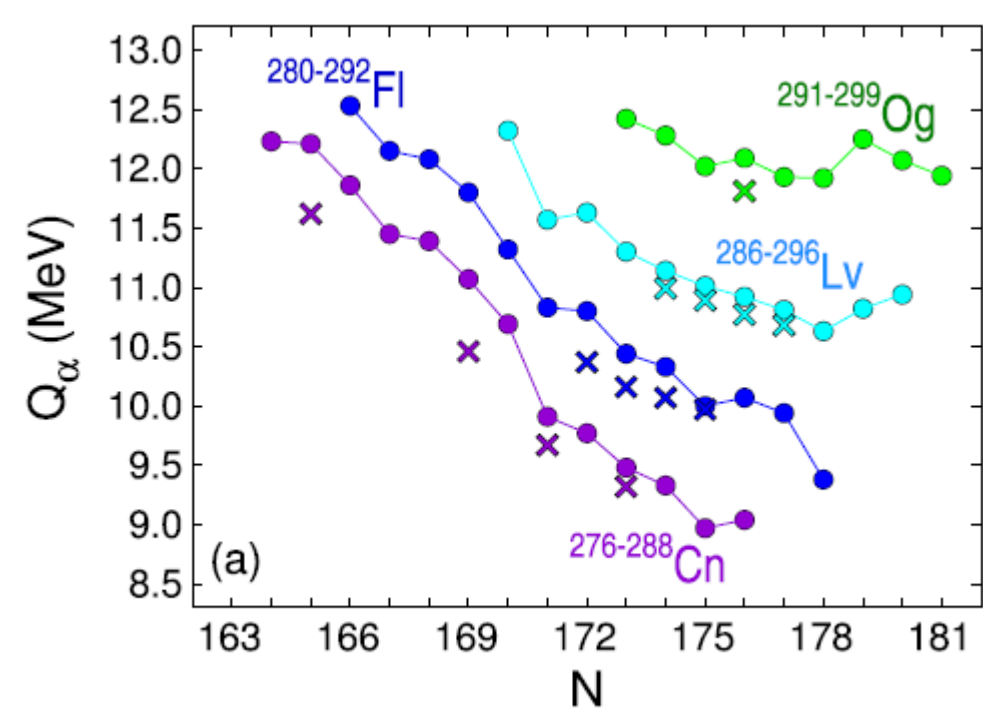
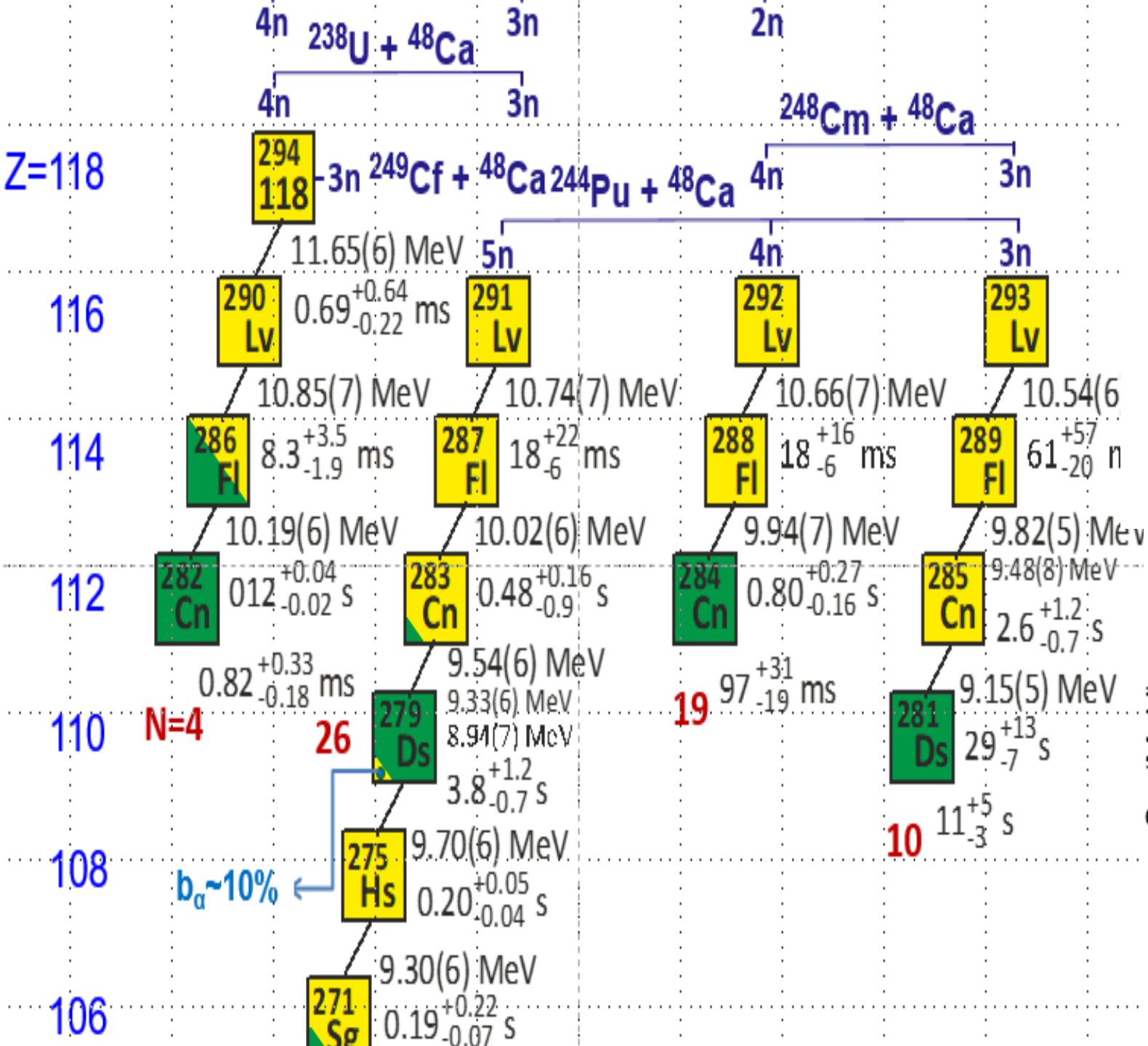


Produced nuclei
lies belong to the
far “island of
stability” of
superheavy
elements.

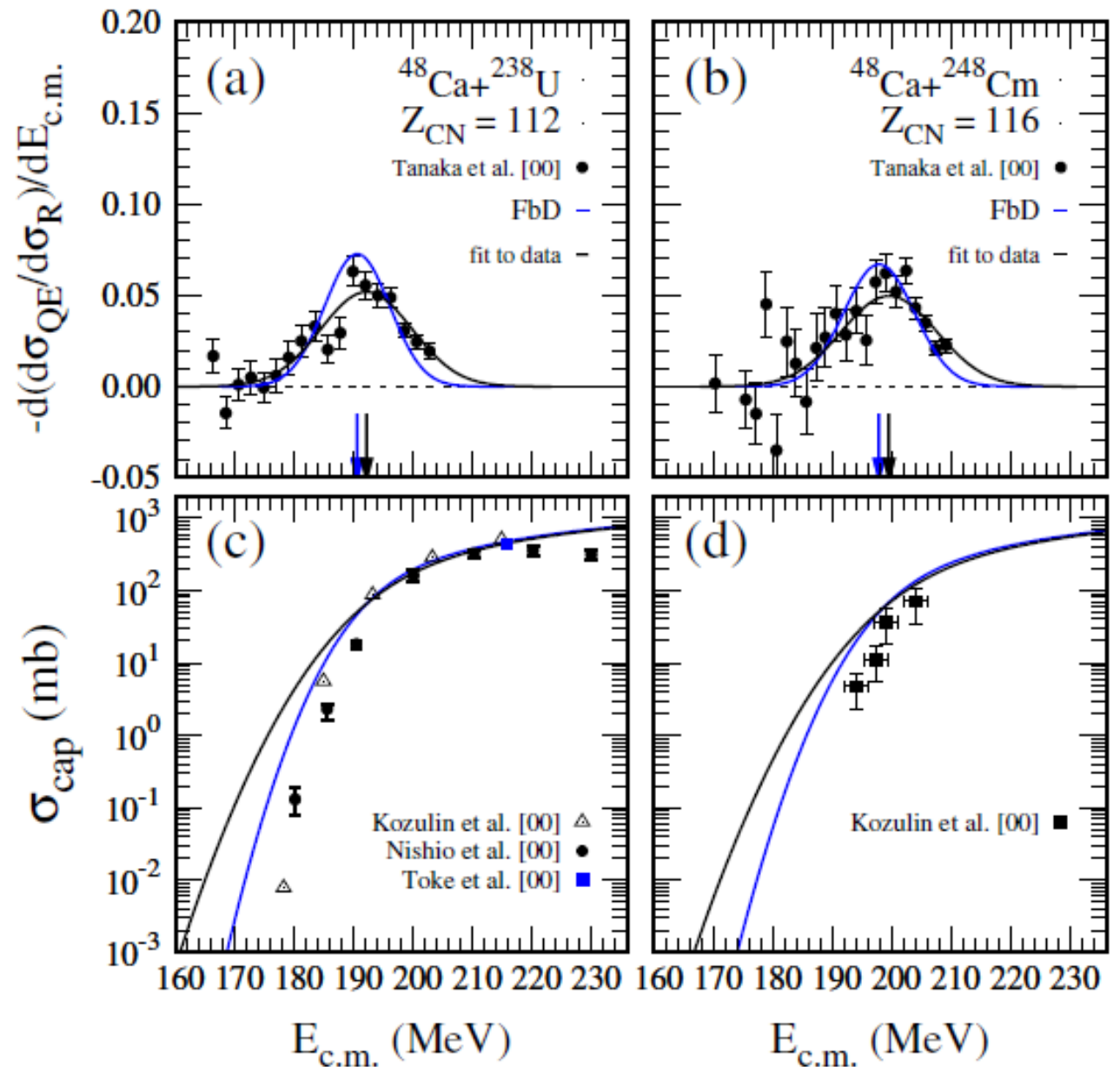
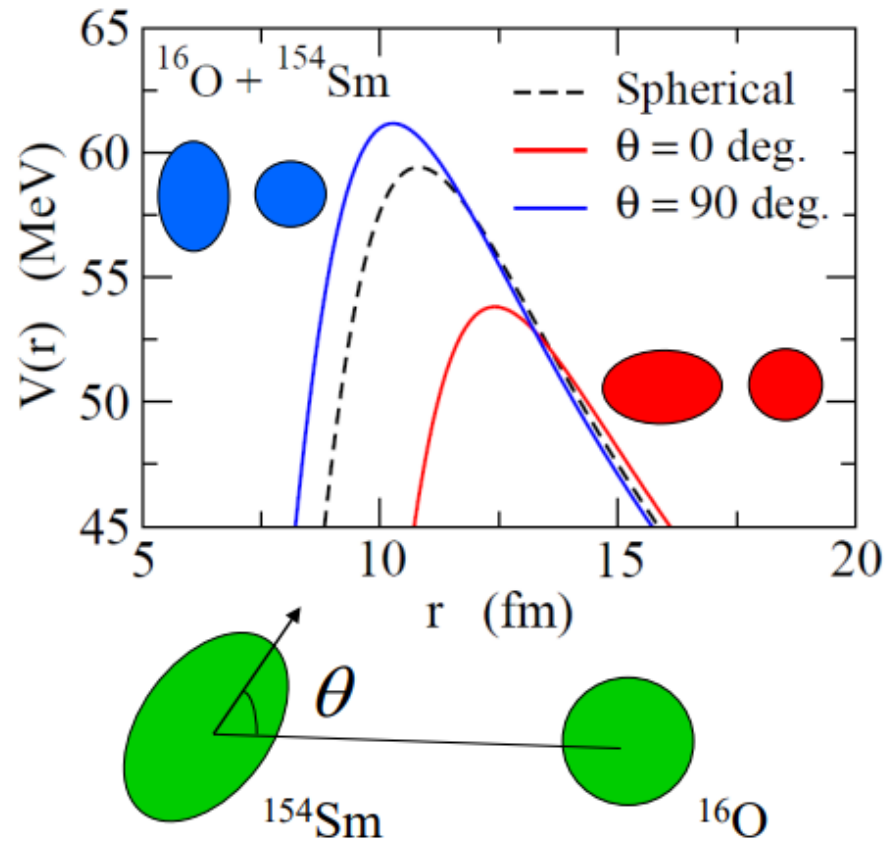


$$\sigma(\text{synthesis}) = \pi \tilde{\lambda}^2 \sum_{l=0}^{l_{\max}} (2l+1) T_l \times P_l(\text{fusion}) \times P_{x_n}^{\ell}(\text{survive})$$





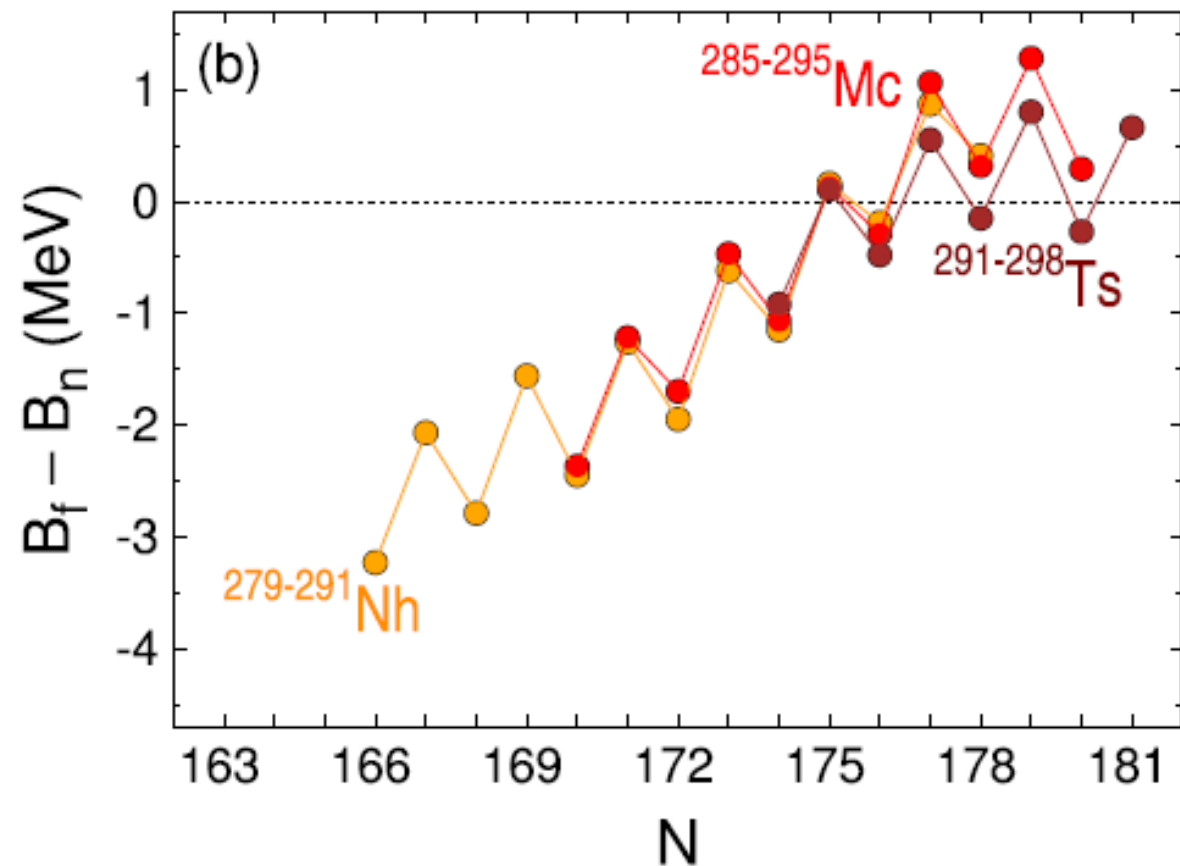
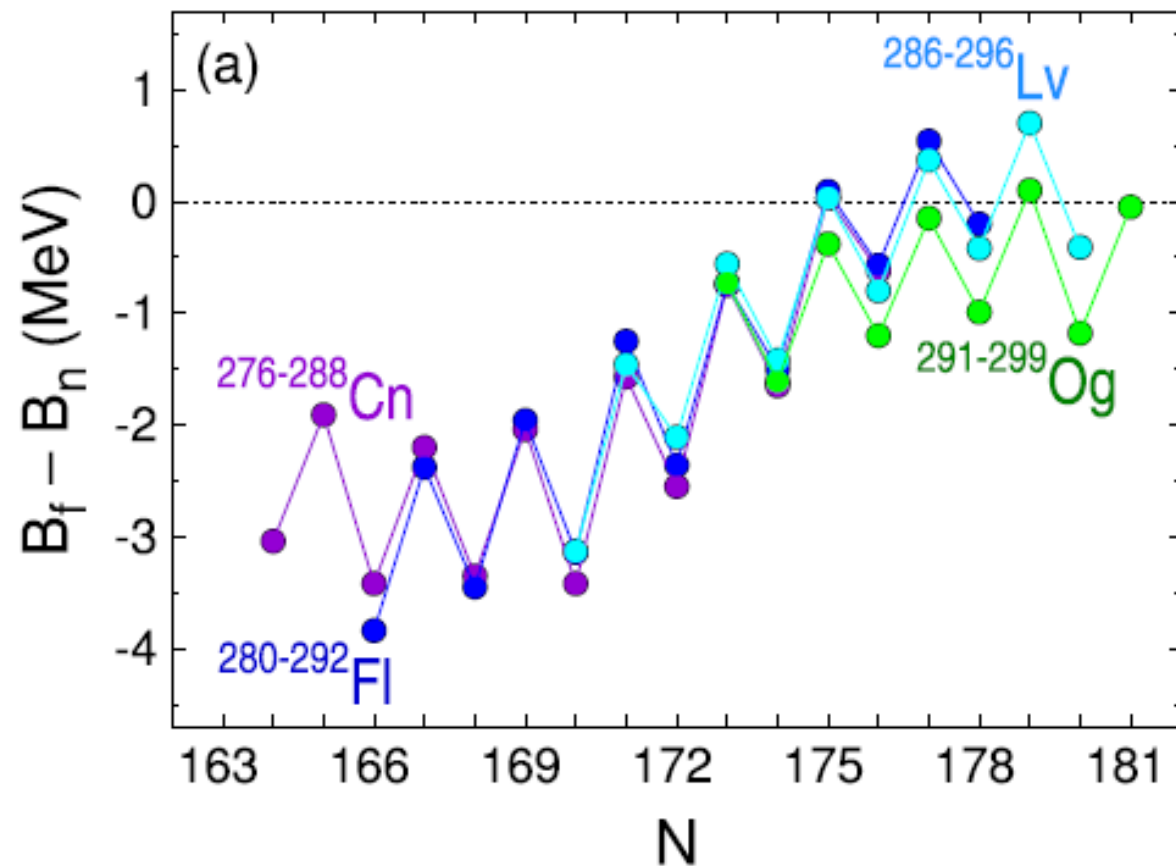
CAPTURE



T. Cap, M. Kowal,, K. Siwek-Wilczynska, Eur. Phys. J. A
In preparation

SURVIVAL

$$\frac{\Gamma_n}{\Gamma_f} = \frac{4mR_0^2}{\hbar^2} \frac{\int_0^{E^*-B_n} \epsilon \rho(E^* - B_n - \epsilon) d\epsilon}{\int_0^{E^*-B_f} \rho(E^* - B_f - K) dK}, \quad E^* > B_n > B_f > T, \quad \longrightarrow \quad \frac{\Gamma_n}{\Gamma_f} = \frac{4mR_0^2}{\hbar^2} \exp\left(-\frac{B_n - B_f}{T}\right).$$



Letter

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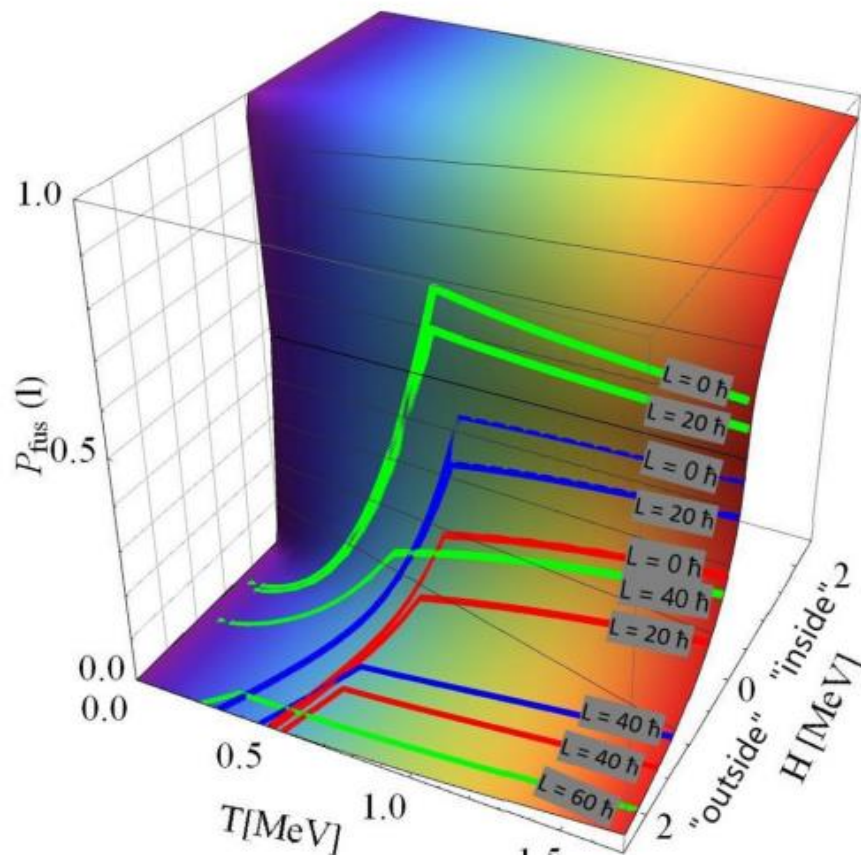
Diffusion as a possible mechanism controlling the production of superheavy nuclei in cold fusion reactions

T. Cap, M. Kowal, and K. Siwek-Wilczyńska
 Phys. Rev. C **105**, L051601 – Published 16 May 2022



Langevin in the overdamped limit:

$$P_{\text{CN}}(E) = \frac{1}{2} \left[1 - \text{erf} \left(\frac{\Delta V}{T} \right) \right]$$



Level-density parameters in superheavy nuclei

A. Rahmatinejad, A. N. Bezbakh, T. M. Shneidman, G. Adamian, N. V. Antonenko, P. Jachimowicz, and M. Kowal

Phys. Rev. C **103**, 034309 – Published 9 March 2021



Energy dependent ratios of level-density parameters in superheavy nuclei

A. Rahmatinejad, T. M. Shneidman, G. G. Adamian, N. V. Antonenko, P. Jachimowicz, and M. Kowal

Phys. Rev. C **105**, 044328 – Published 28 April 2022



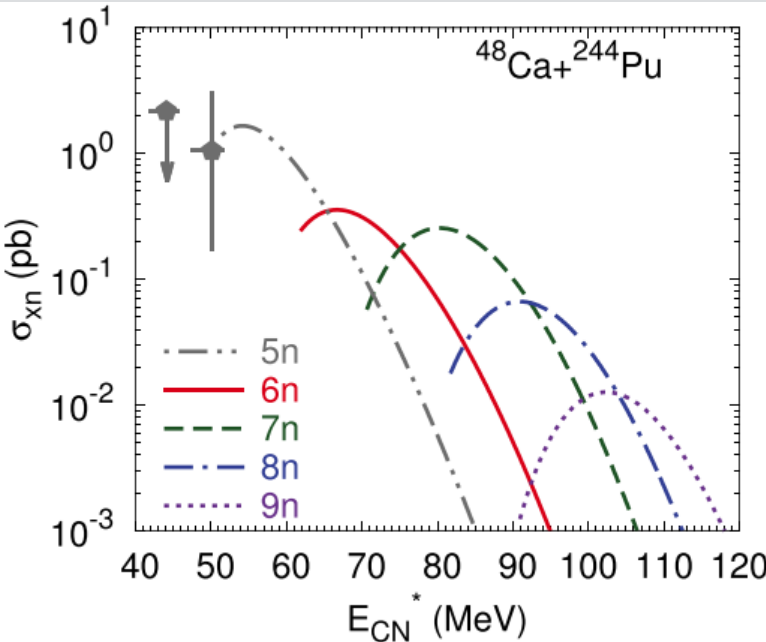
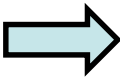
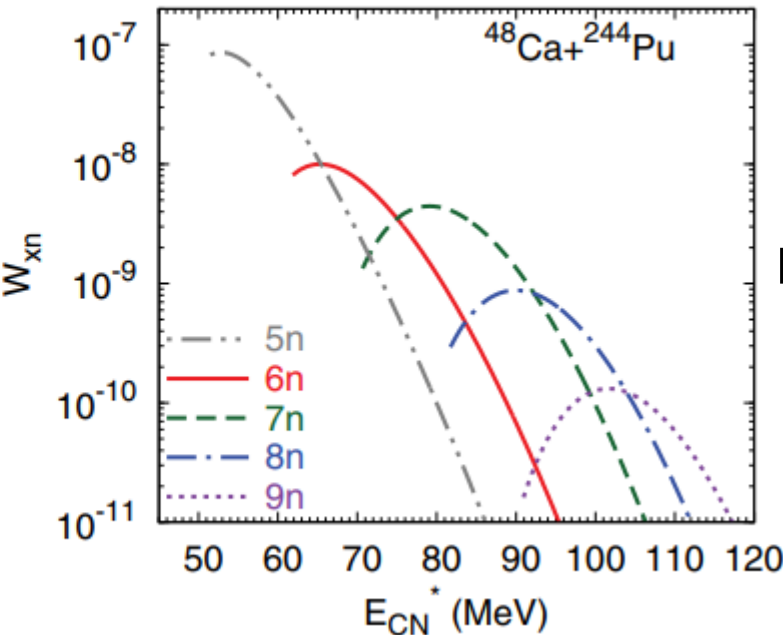
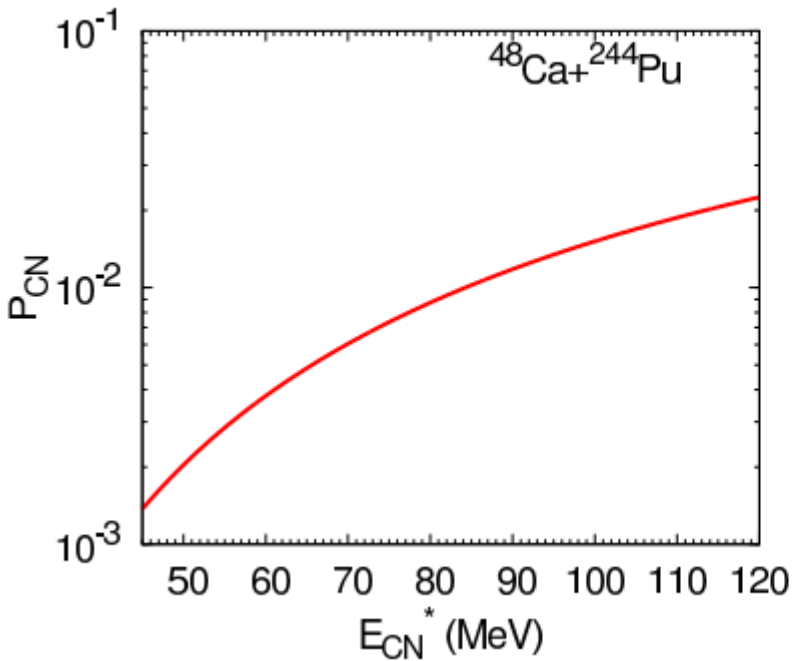
Letter

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Rate of decline of the production cross section of superheavy nuclei with $Z = 114-117$ at high excitation energies

J. Hong, G. G. Adamian, N. V. Antonenko, P. Jachimowicz, and M. Kowal
Phys. Rev. C **103**, L041601 – Published 20 April 2021

the decline of the cross section with increasing excitation energy unexpectedly turned out to be relatively weak. This intriguing behavior may open up a new window for the study and the production of new isotopes



Possibilities of direct production of superheavy nuclei with $Z=112-118$ in different evaporation channels



J. Hong^a, G.G. Adamian^{b,*}, N.V. Antonenko^{b,c}, P. Jachimowicz^d, M. Kowal^e

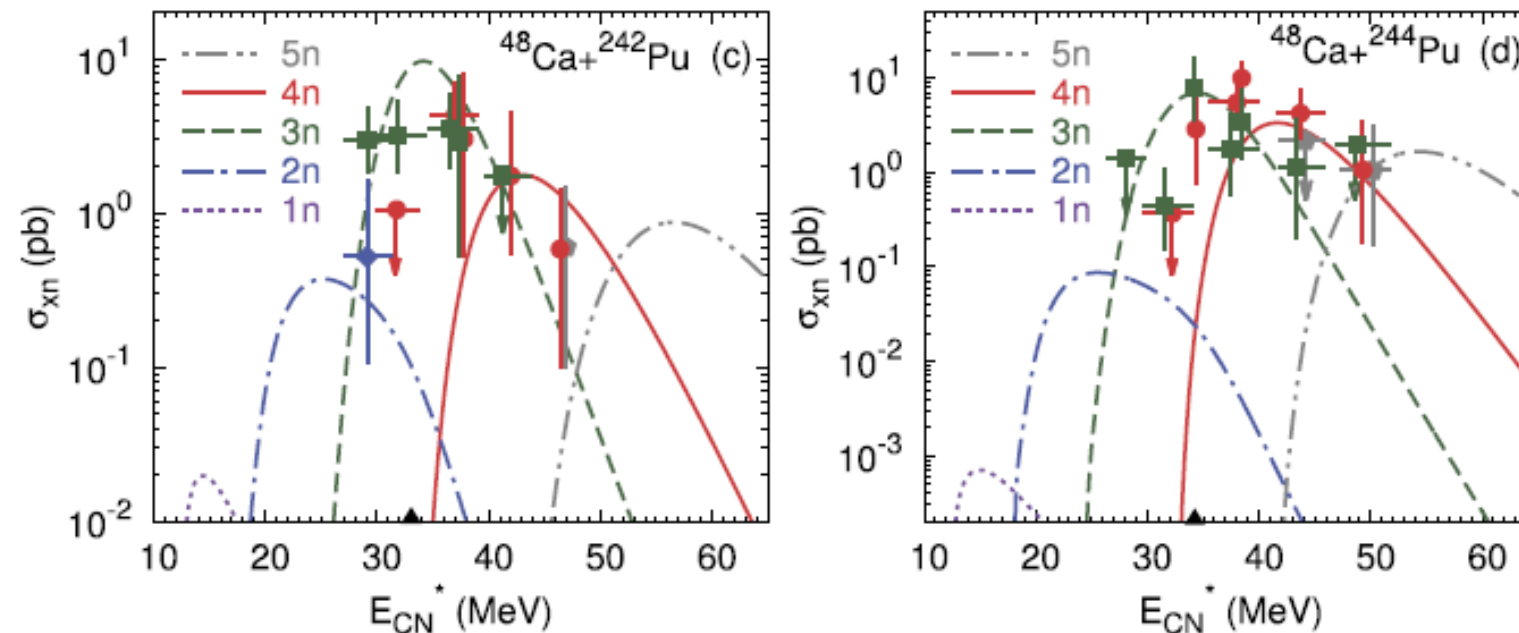
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^e National Centre for Nuclear Research, Pasteura 7, 02-093 Warsaw, Poland



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IUPAC IS NAMING THE FOUR NEW ELEMENTS NIHONIUM, MOSCOVIUM, TENNESSINE, AND OGANESSON

8 June 2016

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IUPAC is naming the four new elements nihonium, moscovium, tennessine, and oganesson

Following [earlier reports](#) that the claims for discovery of these elements have been fulfilled [1, 2], the discoverers have been invited to propose names and the following are now disclosed for public review:

Nihonium and symbol Nh, for the element 113,
Moscovium and symbol Mc, for the element 115,
Tennessine and symbol Ts, for the element 117, and
Oganesson and symbol Og, for the element 118.

The [IUPAC Inorganic Chemistry Division](#) has reviewed and considered these proposals and recommends these for acceptance. A five-month public review is now set, expiring 8 November 2016, prior to the formal approval by the IUPAC Council.

IUPAC Periodic Table of the Elements

1												18											
1 H hydrogen 1.0080 ± 0.0002												2 He helium 4.0026 ± 0.0001											
2																							
3 Li lithium 6.94 ± 0.06												4 Be beryllium 9.0122 ± 0.0001											
11 Na sodium 22.990 ± 0.001												12 Mg magnesium 24.305 ± 0.002											
3												4											
5												6											
7												8											
9												10											
11												12											
13 B boron 10.81 ± 0.02												14 C carbon 12.011 ± 0.002											
15 N nitrogen 14.007 ± 0.001												16 O oxygen 15.999 ± 0.001											
17 F fluorine 18.998 ± 0.001												10 Ne neon 20.180 ± 0.001											
13 Al aluminium 26.982 ± 0.001												14 Si silicon 28.085 ± 0.001											
15 P phosphorus 30.974 ± 0.001												16 S sulfur 32.06 ± 0.02											
17 Cl chlorine 35.45 ± 0.01												18 Ar argon 39.95 ± 0.16											
19 K potassium 39.098 ± 0.001												20 Ca calcium 40.078 ± 0.004											
21 Sc scandium 44.956 ± 0.001												22 Ti titanium 47.867 ± 0.001											
23 V vanadium 50.942 ± 0.001												24 Cr chromium 51.996 ± 0.001											
25 Mn manganese 54.938 ± 0.001												26 Fe iron 55.845 ± 0.002											
27 Co cobalt 58.933 ± 0.001												28 Ni nickel 58.693 ± 0.001											
29 Cu copper 63.546 ± 0.003												30 Zn zinc 65.38 ± 0.02											
31 Ga gallium 69.723 ± 0.001												32 Ge germanium 72.630 ± 0.008											
33 As arsenic 74.922 ± 0.001												34 Se selenium 78.971 ± 0.008											
35 Br bromine 79.904 ± 0.003												36 Kr krypton 83.798 ± 0.002											
37 Rb rubidium 85.468 ± 0.001												38 Sr strontium 87.62 ± 0.01											
39 Y yttrium 88.906 ± 0.001												40 Zr zirconium 91.224 ± 0.002											
41 Nb niobium 92.906 ± 0.001												42 Mo molybdenum 95.95 ± 0.01											
43 Tc technetium [97]												44 Ru ruthenium 101.07 ± 0.02											
45 Rh rhodium 102.91 ± 0.01												46 Pd palladium 106.42 ± 0.01											
47 Ag silver 107.87 ± 0.01												48 Cd cadmium 112.41 ± 0.01											
49 In indium 114.82 ± 0.01												50 Sn tin 118.71 ± 0.01											
51 Sb antimony 121.76 ± 0.01												52 Te tellurium 127.60 ± 0.03											
53 I iodine 126.90 ± 0.01												54 Xe xenon 131.29 ± 0.01											
55 Cs caesium 132.91 ± 0.01												56 Ba barium 137.33 ± 0.01											
57-71 lanthanoids												72 Hf hafnium 178.49 ± 0.01											
73 Ta tantalum 180.95 ± 0.01												74 W tungsten 183.84 ± 0.01											
75 Re rhenium 186.21 ± 0.01												76 Os osmium 190.23 ± 0.03											
77 Ir iridium 192.22 ± 0.01												78 Pt platinum 195.08 ± 0.02											
79 Au gold 196.97 ± 0.01												80 Hg mercury 200.59 ± 0.01											
81 Tl thallium 204.38 ± 0.01												82 Pb lead 207.2 ± 1.1											
83 Bi bismuth 208.98 ± 0.01												84 Po polonium [209]											
85 At astatine [210]												86 Rn radon [222]											
87 Fr francium [223]												88 Ra radium [226]											
89-103 actinoids												104 Rf rutherfordium [267]											
105 Db dubnium [268]												106 Sg seaborgium [269]											
107 Bh bohrium [270]												108 Hs hassium [269]											
109 Mt meitnerium [277]												110 Ds darmstadtium [281]											
111 Rg roentgenium [282]												112 Cn copernicium [285]											
113 Nh nihonium [286]												114 Fl flerovium [290]											
115 Mc moscovium [290]												116 Lv livermorium [293]											
117 Ts tennessine [294]												118 Og oganesson [294]											

Key:
atomic number
Symbol
name
abridged standard
atomic weight



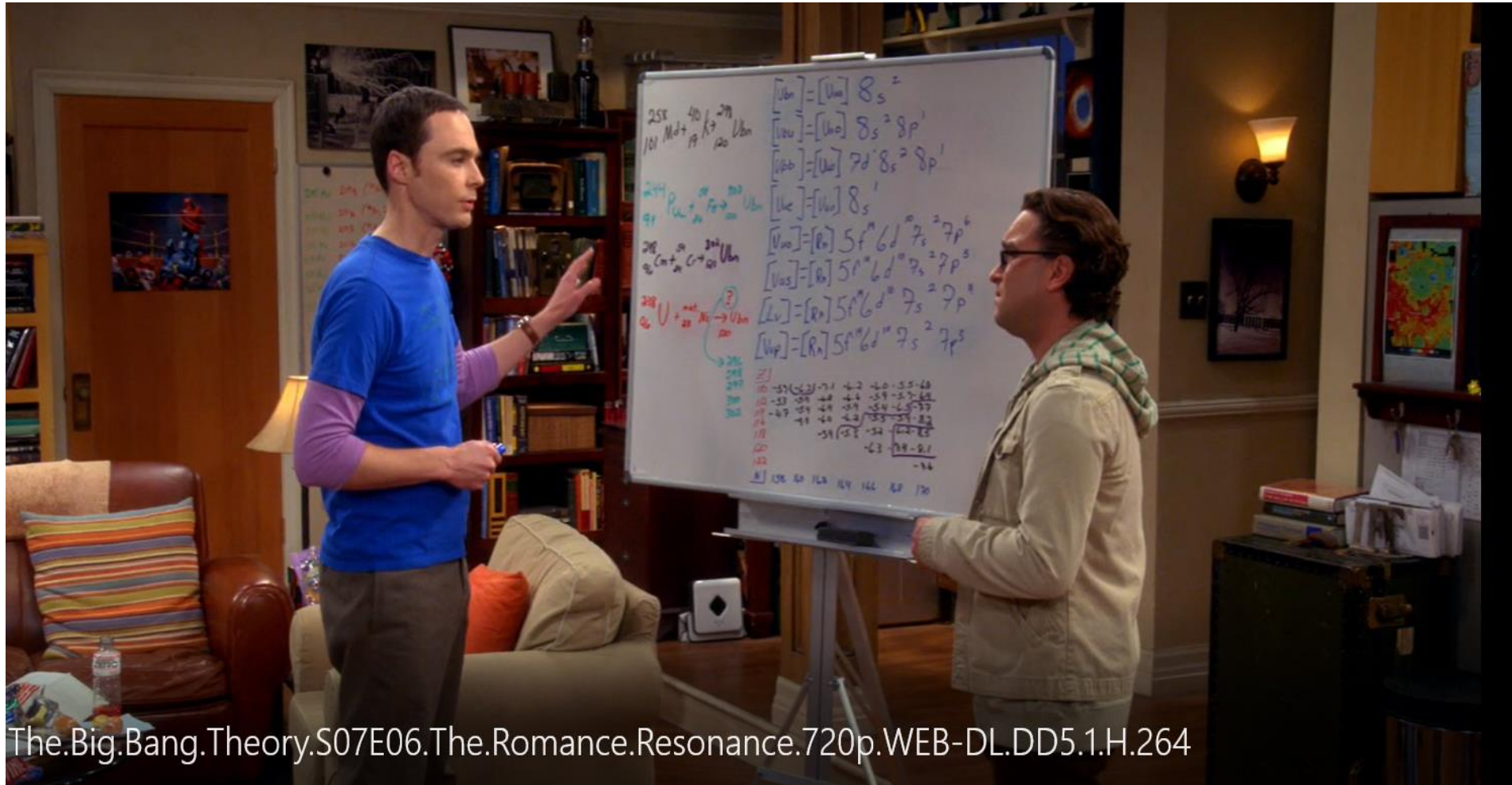
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57 La lanthanum 138.91 ± 0.01	58 Ce cerium 140.12 ± 0.01	59 Pr praseodymium 140.91 ± 0.01	60 Nd neodymium 144.24 ± 0.01	61 Pm promethium [145]	62 Sm samarium 150.36 ± 0.02	63 Eu europium 151.96 ± 0.01	64 Gd gadolinium 157.25 ± 0.03	65 Tb terbium 158.93 ± 0.01	66 Dy dysprosium 162.50 ± 0.01	67 Ho holmium 164.93 ± 0.01	68 Er erbium 167.26 ± 0.01	69 Tm thulium 168.93 ± 0.01	70 Yb ytterbium 173.05 ± 0.02	71 Lu lutetium 174.97 ± 0.01
89 Ac actinium [227]	90 Th thorium 232.04 ± 0.01	91 Pa protactinium 231.04 ± 0.01	92 U uranium 238.03 ± 0.01	93 Np neptunium [237]	94 Pu plutonium [244]	95 Am americium [243]	96 Cm curium [247]	97 Bk berkelium [247]	98 Cf californium [251]	99 Es einsteinium [252]	100 Fm fermium [257]	101 Md mendelevium [258]	102 No nobelium [259]	103 Lr lawrencium [262]

For notes and updates to this table, see www.iupac.org. This version is dated 4 May 2022.

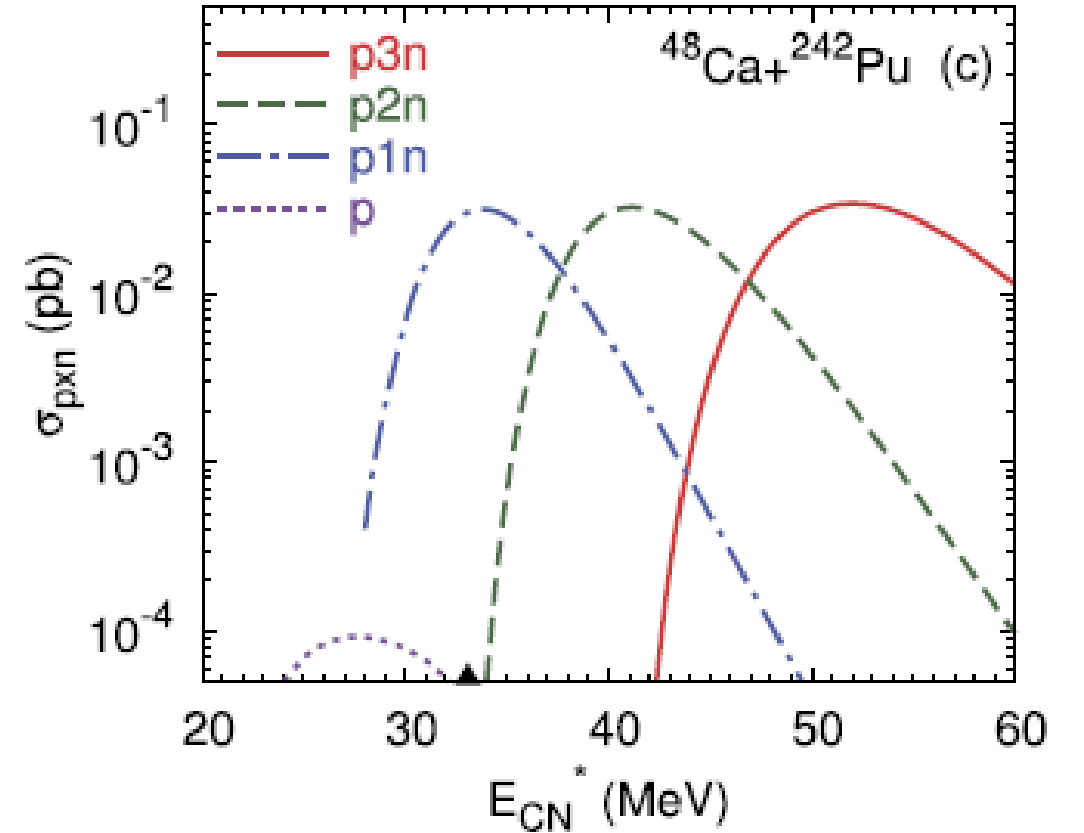
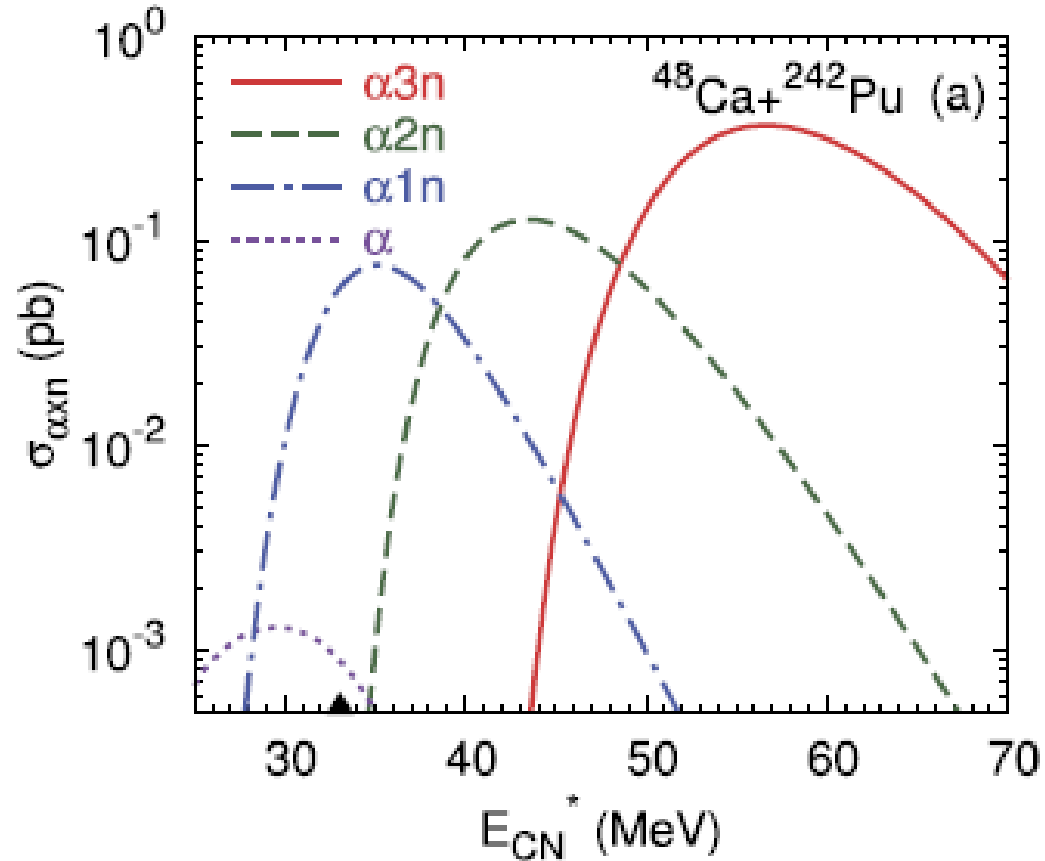
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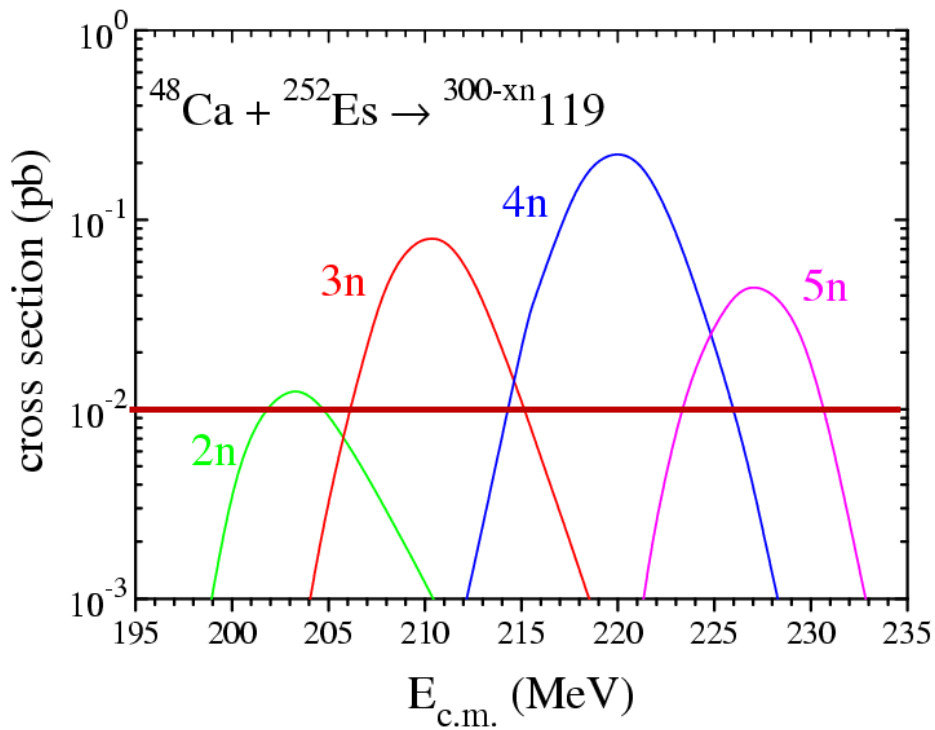
What Next ?



The.Big.Bang.Theory.S07E06.The.Romance.Resonance.720p.WEB-DL.DD5.1.H.264

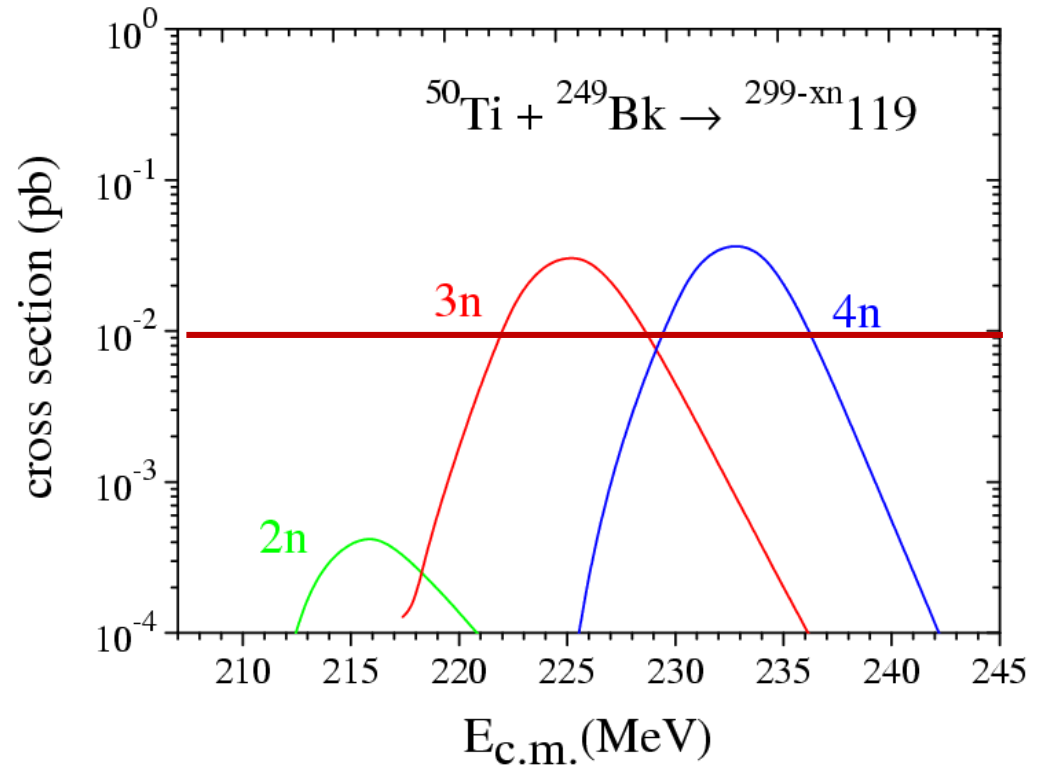
The pxn - and αxn -evaporation channels allow us to obtain an access to the isotopes which are unreachable in xn -evaporation channels due to the lack of proper projectile-target combination. Thus, employing the reactions suggested, one can produce the heaviest isotopes closer to the center of the island of stability.



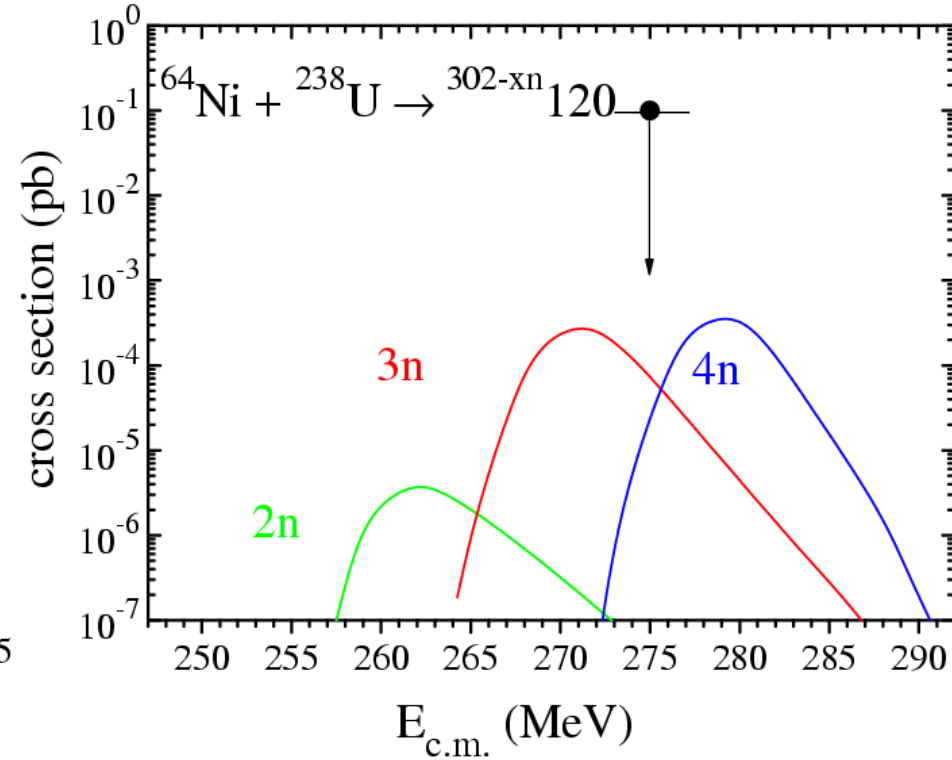
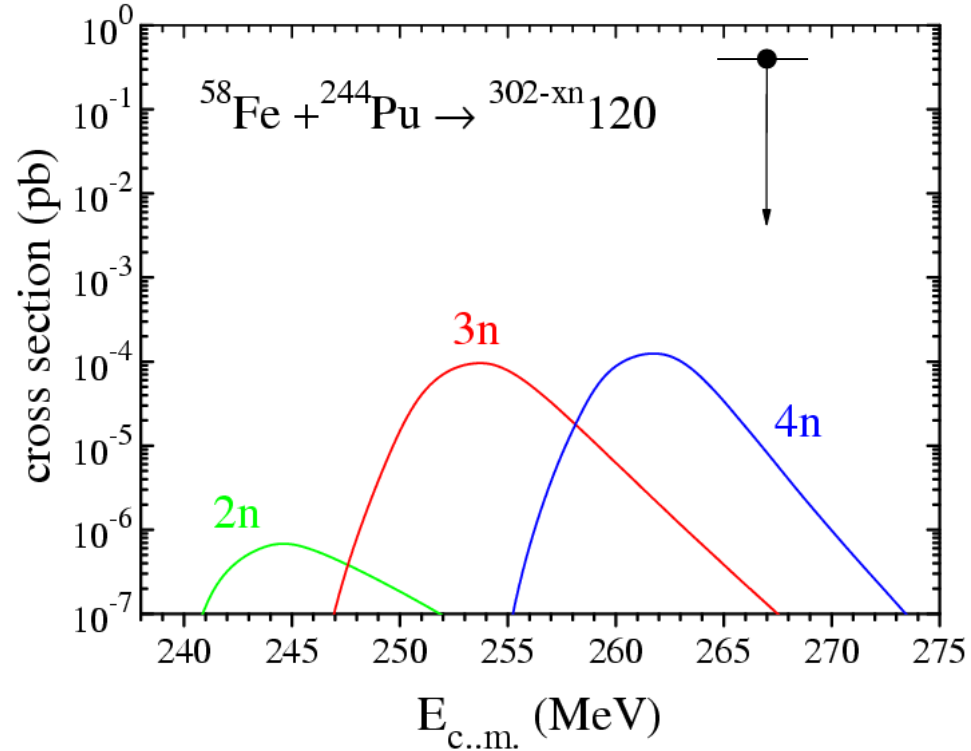


The reaction $^{48}\text{Ca} + ^{252}\text{Es}$ predicted to have measurable cross section

The systematics of the s_{inj} used to predict cross sections for $Z = 119$



$Z = 120$



These values of the evaporation residue cross sections are much smaller (at least one order of magnitude) than previously published predictions.

Thank you for your attention
&
for the prize



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